Concise synthesis of (\pm) -horsfiline and (\pm) -coerulescine by tandem cyclisation of iodoaryl alkenyl azides \dagger

Dimitrios E. Lizos and John A. Murphy*

Department of Pure and Applied Chemistry, University of Strathclyde, 295 Cathedral Street, Glasgow, Scotland, UK Gl 1XL. E-mail: John.murphy@strath.ac.uk

Received 19th August 2002, Accepted 5th September 2002 First published as an Advance Article on the web 29th November 2002

A brief, efficient and economical synthesis of the spiropyrrolidinyloxindoles, horsfiline and coerulescine, has been achieved, starting from itaconic acid and, respectively, *p*-anisidine or *o*-iodoaniline. Tandem radical cyclisation of iodoaryl alkenyl azides is the key step in both syntheses.

Introduction

The recent isolation of the natural products ¹ spirotryprostatin A 1 and spirotryprostatin B 2 (Scheme 1), and the discovery of their activity (albeit mild activity) as cell cycle inhibitors has added considerable impetus to research on the spiropyrrolidinyloxindole alkaloids. Total syntheses ² of the spirotryprostatins A and B have been achieved, but it transpires that synthetic intermediates prepared during Danishefsky's synthetic studies ³ on these compounds, may be more interesting than the spirotryprostatins themselves. Three compounds, 3–5,

† Electronic supplementary information (ESI) available: ¹H and ¹³C NMR spectra are provided for the compounds; the preparations of compounds 19, 20 and 21 and the conversion of 20 to 16 are also included. See http://www.rsc.org/suppdata/ob/b2/b208114h/

were identified to have potent and selective action on human breast cancer cell lines.

Among these, tricyclic compound 3 is exciting since it is a relatively uncomplicated potential lead for development of novel anti-tumour agents. Also highly active was the pair of diastereomers 4 and 5. The discovery of equal activity for the diastereomeric pair shows that the stereochemistry of the spiro centre in these molecules is not crucial for anti-tumour action. Although the cellular site of action of these three compounds has not yet been disclosed, if it is assumed that they act at a common locus, as a working hypothesis, then the anti-tumour activity of tricyclic spiropyrrolidinyloxindoles requires further investigation, and therefore synthetic routes that provide easy access to these classes of compounds become important.

It is not surprising therefore that there has been a significant and increasing interest in the synthesis of the tricyclic natural

1, Spirotryprostatin A

2, Spirotryprostatin B

3

6, R = OMe, horsfiline **7**, R = H, coerulescine

Scheme 1

products, horsfiline 4a-h,4j 6 (isolated by Bodo and co-workers 4a) and coerulescine 4d,i-1 7 (isolated 5 by Colegate and co-workers) that bear the same tricyclic skeleton as 3. A range of synthetic approaches has been used. With respect to horsfiline, Jones and Wilkinson 4b used a free radical cyclisation as the key step in their synthesis. Lévy and co-workers produced 4d two routes featuring (i) oxidative rearrangement of a tetrahydrocarboline (see also ref. 4a) and (ii) oxidation and formaldehyde-treatment of an appropriately substituted tryptamine. In a synthesis of (-)horsfiline as well as, separately, the unnatural enantiomer, Borschberg and co-workers 4c studied the stereospecific conversion of tryptamine derivatives firstly to tetrahydrocarbolines and thence to spiropyrrolidinylindolones, while Palmisano et al. prepared 4e (-)-6 via the (chiral auxiliary-directed) cycloaddition of azomethine ylides onto 3-methyleneindolones. Meanwhile, Fuji and co-workers used 4f asymmetric nitroolefination of indolones to introduce the correct stereochemistry at what would become the spiro-carbon, subsequently making the pyrrolidine ring. More recently, in preparations of (\pm) -6, Carreira's group 4g have ring-expanded a spiro-cyclopropylindolone to the requisite spiropyrrolidine, and Kumar et al. rearranged 4 an (aziridinoethylidene)indolone to a pyrrolidinylindolone.

From our perspective, the spiropyrrolidinyl-oxindole skeleton should be an ideal target for synthesis using the iodoaryl azide tandem radical cyclisation strategy that we recently exploited in the total synthesis of aspidospermidine ⁶ and in the formal total synthesis of vindoline. ⁷ Thus tandem cyclisation of a radical, **8**, should afford the tricyclic system, **9**, in a single process. This paper reports ⁸ the realisation of that goal.

Results and discussion

To synthesise horsfiline, dilithiated t-BOC-p-anisidine 10a was iodinated with 1,2-diiodoethane, to give 11a (86%). Deprotection of the t-BOC group gave 12a (89%), which was reductively alkylated with benzaldehyde 10 (94%). (Conversion of the primary amine 12a to a secondary amine such as 13a is necessary since the radical cyclisations about to be undertaken on the amide substrates, e.g. 17 in Scheme 2 do not succeed 4b with secondary amides. We chose benzylation for this conversion, because of the later easy removal of the benzyl group). Condensation of the benzylamine 13a with the acid chloride 11 14 of the monomethyl ester of itaconic ± acid (in turn readily available 11 from commercially available itaconic acid) afforded the amide 15a (94%). It was now necessary to reduce the ester group in amide 15a to an alcohol, 16a, before conversion to azide 17a and cyclisation. Initial attempts to reduce the ester to alcohol 16a with DIBAL-H led to the desired product 16a as the major product, but it was contaminated with a minor compound that, based on examination of the NMR spectrum, we propose to be 18, where unexpected reduction of the alkene has occurred. Unfortunately the two compounds had similar physical properties, so separation was not possible. To overcome this problem, protection of the alkene double bond in 15a was undertaken. Reaction of 15a with PhSeNa, formed in situ by reduction of diphenyl diselenide with sodium borohydride, surprisingly led to no reaction, and we attribute this to steric hindrance to attack on the alkene by the large phenylseleno group. To decrease the steric problems, PhSH was instead reacted with ester 15a in the presence of base to give an excellent yield of product 19 (96%). Reduction of 19 afforded the desired alcohol 20 (55%) together with a small amount of unexpected amine 21 (10%). Alcohol 20 was purified and oxidized to the corresponding sulfoxide, which was directly subjected to thermal elimination affording the alcohol 16a (72%).

Although **16a** had now been produced, the process of protecting and later deprotecting the alkene detracted from the directness of the synthetic route. With this in mind, we returned to investigate further whether a clean reduction of **15a** to **16a** could be found. Gratifyingly, success was achieved with LiBH₄, generated *in situ* from lithium chloride and sodium borohydride. This afforded the desired product cleanly and in high yield (83%). Conversion to the azide **17a** (68%) was accomplished using diphenylphosphoryl azide, and this was followed by tandem cyclisation to afford the tricyclic product **22a**; this was subjected to *in situ* methylation to yield **23a** (60% over two steps). Finally deprotection under Birch conditions afforded horsfiline **6** (87%).

Significantly, the tandem radical cyclisation afforded solely the desired tricyclic skeleton; no product arising from apparent 6-endo-trig cyclisation of the aryl radical onto the alkene was seen. In studies related to the synthesis of these same alkaloids, Jones et al. had shown 13 how the balance of 5-exo to apparent 6-endo cyclisation is determined by kinetic and thermodynamic factors. At lower temperatures, the 5-exo predominates, 14 but at higher temperatures, rearrangement to 6-endo can occur through a neophyl rearrangement. In our case, the product of the initial 5-exo cyclisation is likely to be trapped by the azide group, preventing side-products from forming.

The same protocol was then followed for the synthesis of coerulescine and with generally excellent yields, producing intermediates 13b (92%), 15b (94%), 16b (84%), 17b (70%) and 23b (68%). We were a little concerned about the final deprotection step to afford coerulescine. Debenzylation of 23a in the horsfiline case involves selective addition of solvated electrons to the aromatic ring of the benzyl protecting group. It is entirely reasonable that this should be a very selective process, since the aromatic ring in the horsfiline skeleton is likely to be far more electron-rich, and hence more difficult to reduce, than that of the benzyl group, due to the electron-donating properties of the methoxy and amide groups. With coerulescine, the absence of the methoxy group should make the difference between the two rings less pronounced, but the benzyl group should still bear the more electron-poor ring. In practice, the selectivity was maintained and the reaction proceeded in an excellent (83%) yield.

It is seen that the iodoaryl alkenyl azide tandem radical cyclisation reaction affords a rapid route to (\pm) -horsfiline in 20% yield from t-BOC-p-anisidine [18% from anisidine], and (\pm) -coerulescine in 29% yield from 2-iodoaniline. An acid chloride derived from the economical itaconic acid is the key partner reagent for both syntheses. The efficiency of this route ranks with the highest reported.

In mechanistic terms, this route shows greatest similarities with the route of Jones and Wilkinson, ⁴⁶ since both use radical cyclisations as the key steps (Scheme 3). However, the two routes differ considerably in all other facets, (e.g. single cyclisation versus tandem cyclisation) and most particularly in the nature and reactivity of the final radicals, carbon radical 24 in their case, versus nitrogen radical 25 in this case. Importantly, in designing the synthesis of more complex spiropyrrolidinyl-indolones, the diversity of efficient routes should now allow access to a wide variety of substitution patterns of analogues for anti-tumor testing.

Experimental section

Melting points were carried out on a Reichert hot stage apparatus and are uncorrected. Infra red spectra were obtained on a Perkin Elmer 1720-X FTIR. Mass spectra were recorded on a JEOL JMS-AX505 HA instrument using, electron impact ionisation (EI⁺), chemical ionisation (CI), electrospray (ES) or fast atom bombardment (FAB). Some low and high resolution mass spectra were obtained on JLSX 102 and VG ZAB-E instruments using peak-matching techniques at the EPSRC

[‡] The IUPAC name for itaconic acid is methylenesuccinic acid.

Scheme 2 Syntheses of horsfiline and coerulescine. *Reagents and conditions*: (a) t-BuLi, Et₂O, -20 °C then ICH₂CH₂I, 86% for **11a**; (b) TFA, DCM, 0 °C, 89% for **12a**; (c) ZnCl₂, PhCHO, MeOH, NaCNBH₃, 94% for **13a**, 92% for **13b**; (d) Et₃N, PhH, 94% for **15a**; 94% for **15b**; (e) PhSH, DBU, THF, Δ , 96% for **19**; (f) DIBAL-H, PhMe, -78 °C, 55% for **20**; (g) m-CPBA, DCM, then PhMe, Δ , 72 h, 72% for **16a**; (h) LiCI, NaBH₄, EtOH, THF, 83% for **16a**, 84% for **16b**; (i) PPh₃, (PhO)₂P(O)N₃, DEAD, THF, 68% for **17a**; 70% for **17b**; (j) (Me₃Si)₃SiH, AIBN, PhH; (k) CH₂O, NaCNBH₃, MeCN, yield over 2 steps: 60% for **23a**, 68% for **23b**; (1) Na, NH₃, 86% for **6**, 83% for **7**.

Scheme 3 The different regiochemical outcomes of the two radical cyclisation approaches to spiropyrrolidinylindolones.

Mass Spectrometry Centre, Swansea. 1 H NMR spectra were recorded at 400 MHz on a Bruker DPX 400 spectrometer. 13 C{H} NMR spectra were similarly recorded at 100.61 MHz on a Bruker DPX400 machine. NMR experiments were carried out in deuterochloroform with tetramethylsilane as internal reference unless otherwise stated. 13 C NMR spectra were acquired in a broadband decoupled mode with the multiplicities obtained using a DEPT or Jmod sequence. Chemical shifts (δ) are reported in parts per million (ppm) from tetramethylsilane. The following abbreviations are used for multiplicities: s = singlet; d = doublet; t = triplet; q = quartet; m =

multiplet, br = broad. Coupling constants (J) are reported in Hertz (Hz).

tert-Butyl (2-iodo-4-methoxyphenyl)carbamate 11a9

Addition of *tert*-butyllithium in pentane (1.7 M, 39.6 cm⁻³, 67.2 mmol, 2.5 equiv.) to a solution of *tert*-butyl (4-methoxyphenyl)carbamate (6.0 g, 26.9 mmol, 1.00 equiv.), in dry diethyl ether (60 cm⁻³), was performed under a nitrogen atmosphere at -20 °C, followed by stirring for 3 h at the same temperature. To this mixture was added 1,2-diiodoethane (11.4 g, 40.3 mmol,

1.5 equiv.) in dry diethyl ether (80 cm⁻³) at -78 °C, and the reaction was gradually warmed to room temperature and stirred for 16 h. After addition of a saturated aqueous solution of sodium thiosulfate (200 cm⁻³), the mixture was extracted with diethyl ether. The organic extracts were then washed with brine, dried over anhydrous sodium sulfate and evaporated under reduced pressure. The residue was purified by column chromatography (10: 1 petroleum ether-ethyl acetate) to yield tert-butyl (2-iodo-4-methoxyphenyl)carbamate 11a as colourless prisms (8.1 g, 86%); mp 48–49 °C (lit. 9 mp 49–50 °C); [Found (EI) M⁺ 349.0172. C₁₂ H₁₆INO₃ requires M, 349.0175]. v_{max} (film)/cm⁻¹ 3399, 2976, 2933, 2834, 1725, 1511, 1367, 1250, 1161, 1037; $\delta_{\rm H}$ 1.51 (9H, s, 3 × CH₃), 3.72 (3H, s, CH₃), 6.55 (1H, br s, NH), 6.84–6.87 (1H, dd, J 9.0, 2.8 ArH), 7.28 (1H, d, J 2.8, ArH) 7.76 (1H, d, J 9.0, ArH); $\delta_{\rm C}$ 28.5 (q), 55.9 (q), 80.9 (s), 90.5 (s), 115.1 (d), 122.3 (d), 123.9 (d), 132.6 (s), 153.3 (s), 156.3 (s). m/z (EI) 349 (29%, M⁺), 293 (96), 249 (46), 166 (34), 106 (22) and 57 (100).

4-Amino-3-iodoanisole 12a

Trifluoroacetic acid (21.2 cm⁻³, 274.9 mmol, 12.00 equiv.) was added to a solution of tert-butyI (2-iodo-4-methoxyphenyl)carbamate (8.0 g, 22.9 mmol, 1.00 equiv.) in dry DCM (120 cm⁻³) at 0 °C and the reaction was stirred at room temperature for 2 h. Sodium hydroxide (2 M) was added followed by hydrochloric acid (2 M) until neutral pH had been attained. The two layers were separated and the aqueous one was extracted with DCM. The organic layers were washed with brine, dried over anhydrous sodium sulfate and the solvent was removed in vacuo. The crude product was purified by column chromatography (4: 1 petroleum ether-ethyl acetate) giving 4-amino-3-iodoanisole 12a as a colourless oil (5.1 g, 89%); [Found (ES): MH^+ 249.9732 C_7H_9INO , requires MH, 249.9729]; v_{max} (film)/cm⁻¹ 3431, 3348, 2939, 2830, 1596, 1493, 1232, 1037; $\delta_{\rm H}$ 3.62 (2H, br s, NH₂), 3.77 (3H, s, CH₃), 6.74 (1H, d, J, 8.7, ArH), 6.82 (1H, dd, J 8.7, 2.7, ArH), 7.26 (1H, d, J 2.7, ArH); $\delta_{\rm C}$ 56.4 (q), 84.7 (s), 115.9 (d), 116.5 (d), 124.0 (d), 141.3 (s), 153.1 (s); m/z (CI) 267 (MNH₄⁺, 63%), 250 (M⁺, 100).

Benzyl(2-iodo-4-methoxyphenyl)amine 13a

A stirred mixture of 4-amino-3-iodoanisole 12 (3 g, 12.00 mmol, 1.00 equiv.), zinc(II) chloride (1.97 g, 14.45 mmol, 1.20 equiv.) and benzaldehyde (1.47 cm⁻³, 14.45 mmol, 1.20 equiv.) in methanol, (50 cm⁻³) was treated with sodium cyanoborohydride (0.91 g, 14.45 mmol, 1.20 equiv.) and warmed at reflux under nitrogen for 2 h. The cooled reaction mixture was diluted with sodium hydroxide (1 M, 50 cm⁻³) and extracted with diethyl ether. The combined ether layers were washed with water, dried over anhydrous sodium sulfate and concentrated in vacuo. Benzyl(2-iodo-4-methoxyphenyl)amine 13a was purified by column chromatography on silica gel eluting with petroleum ether-ethyl acetate 5:1 to give pure title compound as a yellow oil (3.85 g, 94%); Found: C, 49.87; H, 4.15; N, 4.06. C₁₄H₁₄INO requires C, 49.56; H, 4.16; N, 4.13%; (Found (ES): MH⁺ 340.0196, $C_{14}H_{15}INO$ requires MH 340.0198); v_{max} (film)/cm⁻¹ 3401, 3028, 2998, 2830, 1603, 1504, 1282, 1039; $\delta_{\rm H}$ 3.75 (3H, s, CH₃), 4.33 (1H, br s, NH), 4.39 (2H, d, J 4.2, CH₂), 6.54 (1H, d, J 8.9, ArH), 6.84, (1H, dd, J 2.9, 8.9, ArH), 7.30–7.42 (6H, m, ArH); $\delta_{\rm C}$ 49.2 (t), 56.1 (q), 85.5 (s), 111.7 (d), 115.7 (d), 124.8 (d), 127.4 (d), 128.8 (d), 139.1 (s), 142.0 (s), 152.2 (s); m/z (CI) 340.1 ([MH]⁺, 100%).

3-Chlorocarbonylbut-3-enoic acid methyl ester 14¹¹

Thionyl chloride (7.1 cm⁻³, 97.2 mmol, 1.40 equiv.) was added to a solution of 2-methylenesuccinic acid 4-methyl ester (10 g, 69.4 mmol, 1.00 equiv.) in dry diethyl ether (8 cm⁻³) and the reaction mixture was refluxed gently until gas evolution was essentially complete (15 min). Distillation gave 3-chloro-

carbonylbut-3-enoic acid methyl ester **14** as a colourless oil (10.8 g, 96%) bp 84–87 °C (11 mmHg) (Found: M^+ 162.0080, $C_6H_7ClO_3$ requires M 162.0084); ν_{max} (film)/cm⁻¹ 3004, 2955, 2847, 1740, 1638, 1437, 1343, 1206; δ_H 3.39 (2H, d, J 0.8, CH₂), 3.72 (3H, s, CH₃), 6.17 (1H, t, J 0.8, CH₂), 6.70 (1H, s, CH₂); δ_C 37.8 (t), 52.8 (q), 136.5 (t), 138.7 (s), 168.6 (s), 170.3 (s); m/z (EI) 163 ([MH] $^+$, 5%), 127 (60), 113 (95), 99 (50), 85 (100).

3-[Benzyl(2-iodo-4-methoxyphenyl)carbamoyl]but-3-enoic acid methyl ester 15a

Triethylamine (1.6 cm⁻³, 11.4 mmol, 1.10 equiv.) was added to a stirred solution of benzyl(2-iodo-4-methoxyphenyl)amine (3.5 g, 10.3 mmol, 1.00 equiv.) in dry benzene (10 cm^{-3}) at room temperature. 3-Chlorocarbonylbut-3-enoic acid methyl ester (2.18 g, 13.4 mmol, 1.30 equiv.) in dry benzene (10 cm^{-3}) was added dropwise. The resulting solution was stirred at rt for 30 min, then heated under reflux for 90 min. After cooling, the solvent was removed under reduced pressure and the residue dissolved in diethyl ether (50 cm⁻³). The ether was washed with water, dried over anhydrous sodium sulfate and concentrated. Column chromatography eluting with petroleum ether-ethyl acetate 4:1 gave pure 3-[benzyl(2-iodo-4-methoxyphenyl)carbamoyl]but-3-enoic acid methyl ester 15a as a white solid mp 107-110 °C (4.5 g, 94%); Found C, 51.68; H, 4.36; N, 3.03. C₂₀H₂₀INO₄ requires C, 51.61; H, 4.33; N, 3.01%; [Found: (ES) MH^{+} 466.0521, $C_{20}H_{21}INO_{4}$ requires MH 466.0515]; v_{max} (film)/ ${\rm cm^{-1}}$ 3065, 2998, 2841, 1736, 1651, 1622, 1491, 1287, 1175; $\delta_{\rm H}$ 3.06 (1H, d, J 16.5, CH₂), 3.64–3.69 (4H, CH₃ + CH₂), 3.78 (3H, s, CH₃), 4.17 (1H, d, J 14.3, CH₂), 5.16 (1H, s, CH₂), 5.27 (1H, s, CH₂), 5.77 (1H, d, J 14.3, CH₂), 6.70 (1H, dd, J 2.8, 8.7, ArH), 6.81 (1H, d, J 8.7, ArH), 7.23-7.31 (5H, m, ArH), 7.42 $(1H, d, J 2.8, ArH); \delta_c 40.1 (t), 52.0 (q), 52.1 (t), 55.7 (q), 100.1$ (s), 114.4 (d), 123.2 (t), 125.0 (d), 127.6 (d), 128.4 (d), 129.6 (d), 132.0 (d), 136.5 (s), 136.8 (s), 137.5 (s), 159.0 (s), 169.2 (s), 171.3 (s); m/z (ES) 466 [(MH)⁺, 60%], 340 (100).

Reduction of 3-[benzyl(2-iodo-4-methoxyphenyl)carbamoyl]but-3-enoic acid methyl ester 15a using NaBH₄ and LiCl

Lithium chloride (16.2 mg, 0.43 mmol, 2.00 equiv.) was added to a solution of 3-[benzyl(2-iodo-4-methoxyphenyl)carbamoyl]but-3-enoic acid methyl ester 15a (100 mg, 0.21 mmol, 1.00 equiv.) in dry THF-EtOH (2:1 mixture, 3 cm⁻³). When it had dissolved, NaBH₄ (16.2 mg, 0.43 mmol, 2.00 equiv.) was added and the reaction mixture was stirred at room temperature for 16 h. Then the solvent was evaporated near dryness and H₂O (4 cm⁻³) was added. The organic phase was extracted with ethyl acetate, washed with H2O, dried over anhydrous sodium sulfate and concentrated under vacuum. The reaction mixture was subjected to column chromatography eluting with petroleum ether-ethyl acetate 1:1 to afford N-benzyl-2-(2hydroxyethyl)-N-(2-iodo-2-methoxyphenyl)acrylamide 16a as a colourless oil (76 mg, 83%); [Found (ES): MH+ 438.0563, $C_{19}H_{21}INO_3$ requires MH 438.0566]; v_{max} (film)/cm⁻¹ 3409, 2938, 1614, 1487, 1287, 1030; $\delta_{\rm H}$ 2.01 (1H, br s, OH) 2.25–2.31 (1H, m, CH₂), 2.48–2.54 (1H, m, CH₂), 3.62–3.77 (5H, m, CH₂) and CH₃), 4.10 (1H, d, J 14.1, CH₂), 5.10 (1H, s, CH₂), 5.14 (1H, s, CH₂), 5.71 (1H, d, J 14.1, CH₂), 6.56 (1H, d, J 8.7 ArH), 6.68 (1H, dd, J 2.7, 8.7, ArH), 7.21–7.39 (5H, m, ArH), 7.40 (1H, d, J 2.7, ArH); δ_C 38.2 (t), 52.4 (t), 55.7 (q), 62.6 (t), 100.1 (s), 114.4 (d), 120.2 (t), 125.0 (d), 127.9 (d), 128.6 (d), 129.6 (d), 131.6 (d), 136.6 (s), 137.1 (s), 142.0 (s), 159.1 (s), 172.2 (s); *m/z* (El) 438.1 [(MH)⁺, 100%]

2-(2-Azidoethyl)-N-benzyl-N-(2-iodo-4-methoxyphenyl)-acrylamide 17a

Triphenylphosphine (363 mg, 1.38 mmol, 1.10 equiv.), DEAD (0.22 cm⁻³, 1.38 mmol, 1.10 equiv.) and diphenylphosphoryl azide (0.3 cm⁻³, 1.38 mmol, 1.10 equiv.) were added sequen-

tially to a solution of alcohol 16a in dry THF (12 cm⁻³), and the solution was stirred at rt for 24 h. Then the solvent was evaporated and the residue was subjected to column chromatography using petroleum ether-ethyl acetate 5:1 to afford pure 2-(2azidoethyl)-N-benzyl-N-(2-iodo-4-methoxyphenyl) acrylamide 17a as a slightly yellow oil (396 mg, 68%); [Found (ES): MH⁺ 463.0622, $C_{19}H_{20}IN_4O_2$ requires MH 463.0631]; v_{max} (film)/cm⁻¹ 3030, 2938, 2838, 2096, 1651, 1622, 1488, 1287, 1030; $\delta_{\rm H}$ 2.34– 2.41 (1H, m, CH₂), 2.55–2.62 (1H, m, CH₂), 3.43 (2H, t, J 6.9, CH₂), 3.78 (3H, s, CH₃), 4.09 (1H, d, J 14.1, CH₂), 5.12 (1H, s, CH₂), 5.17 (1H, s, CH₂), 5.71 (1H, d, J 14.1, CH₂), 6.55 (1H, d, J 8.7, ArH), 6.69 (1H, dd, J 2.7, 8.7, ArH), 7.21–7.35 (5H, m, ArH), 7.41 (1H, d, J 2.7, ArH); δ_C 33.9 (t), 50.1 (t), 52.3 (t), 55.8 (q), 100.4 (s), 114.5 (d), 120.3 (t), 125.1 (d), 127.8 (d), 128.6 (d), 129.7 (d), 131.7 (d), 136.9 (s), 137.1 (s), 140.7 (s), 159.2 (s), 170.5 (s); m/z (CI) 463 [(MH)⁺, 35%], 115 (100).

Tandem radical cyclisation of 2-(2-azidoethyl)-*N*-benzyl-*N*-(2-iodo-4-methoxyphenyl)acrylamide 17a

2-(2-Azidoethyl)-N-benzyl-N-(2-iodo-4-methoxyphenyl)acrylamide 17a (295 mg, 0.64 mmol, 1.00 equiv.) was dissolved in dry benzene (20 cm⁻³) and the solution was purged with argon for 30 min. The solution was brought to reflux and tristrimethylsilylsilane (TTMSS) (0.22 cm⁻³, 0.70 mmol, 1.10 equiv.) and AIBN (21 mg, 0.13 mmol, 0.2 equiv.) were added and the mixture was refluxed for 5 h. The reaction mixture was allowed to cool down to room temperature and the solvent was removed in vacuo. Ethyl acetate (50 cm⁻³) was added and the mixture extracted with hydrochloric acid (2 M). The aqueous phase was basified with sodium hydroxide (2 M) and then it was extracted with ethyl acetate. The organic phase was dried over anhydrous sodium sulfate and concentrated in vacuo. The crude 1-benzyl-5-methoxy-1*H*-spiro[indole-3,3'-pyrrolidin]-2-one **22a** (150 mg, 0.49 mmol, 1.00 equiv.) from the previous step was dissolved in acetonitrile (7 cm⁻³) and treated ^{5f} with formaldehyde [(0.24 cm⁻³, 2.92 mmol, 6.00 equiv.) 37% in water] and NaCNBH₃ (61 mg, 0.97 mmol, 2.00 equiv.) at rt. The mixture was stirred for 15 min and neutralised with acetic acid. After stirring was continued for another 1 h, the mixture was basified with ammonia solution and the solvent was removed in vacuo. The residue was extracted with ethyl acetate. The combined organic extracts were washed with brine, dried over anhydrous sodium sulfate and concentrated under reduced pressure. Column chromatography eluting with ethyl acetate: acetone 4 : 1 gave pure 1-benzyl-5-methoxy-1'-methyl-1H-spiro[indole-3,3'-pyrrolidin]-2-one 23a as a colourless oil (123 mg, 60% over two steps) (Found MH⁺ 323.1761, $C_{20}H_{23}N_2O_2$ requires *MH* 323.1759); v_{max} (film)/cm⁻¹ 2941, 2836, 2787, 1707, 1601, 1491, $1177, 1035; \delta_{H} 2.12-2.20 (1H, m, CH_{2}), 2.40-2.46 (1H, m, CH_{2}),$ 2.51 (3H, s, CH₃), 2.78-2.85 (1H, m, CH₂), 2.91 (1H, d, J 9.4, CH₂), 3.01 (1H, d, J 9.4, CH₂), 3.15–3.20 (1H, m, CH₂), 3.77 (3H, s, CH₃), 4.88 (2H, s, CH₂), 6.59 (1H, d, J 8.5, ArH), 6.68 (1H, dd, J 2.6, 8.5, ArH), 7.11 (1H, d, J 2.6, ArH), 7.24–7.33 $(5H, m, ArH); \delta_C 38.4 (t), 42.0 (q), 44.1 (t), 53.9 (s), 56.1 (q),$ 56.8 (t), 66.2 (t), 109.3 (d), 110.6 (d), 112.5 (d), 127.4 (d), 127.8 (d), 129.0 (d), 135.6 (s), 136.2 (s), 137.0 (s), 156.6 (s), 180.2 (s); m/z (CI) 323 [(MH)⁺, 100%].

(±)-Horsfiline 64g

Ammonia ($ca. 5 \text{ cm}^{-3}$) was condensed at -50 °C and sodium (31 mg) was added with vigorous stirring. 1-Benzyl-5-methoxy-1'-methyl-1H-spiro[indole-3,3'-pyrrolidin]-2-one **23a** (29.4 mg, 0.09 mmol, 1.00 equiv.) was added in THF (1 cm⁻³). After 12 min, a small amount of NH₄Cl was added and the ammonia left to evaporate. Water (25 cm⁻³) was added and the reaction was extracted with ethyl acetate. The organic layer was dried over anhydrous sodium sulfate and the solvent was evaporated *in vacuo*. Column chromatography on silica gel with DCM–

MeOH 9 : 1 afforded pure (±)-horsfiline as a white solid (18.2 mg, 87%); mp 154–156 °C (lit. 4g mp 156–157 °C) [Found (ES): MH+ 233.1287, C₁₃H₁₇N₂O₂ requires MH 233.1290]; $\nu_{\rm max}$ (film)/cm $^{-1}$ 3400, 2944, 2849, 1704, 1604, 1491, 1304, 1209, 1031; $\delta_{\rm H}$ 2.05–2.18 (1H, m, CH₂), 2.38–2.44 (1H, m, CH₂), 2.49 (3H, s, CH₃), 2.77–2.84 (1H, m, CH₂), 2.90 (1H, d, J 9.4, CH₂), 2.97 (1H, d, J 9.4, CH₂), 3.06–3.11 (1H, m, CH₂), 3.80 (3H, s, CH₃), 6.72 (1H, dd, J 2.5, 8.4, ArH), 6.80 (1H, d, J 8.4, ArH), 7.06 (1H, d, J 2.5, ArH), 8.32 (1H, br s, NH); $\delta_{\rm C}$ 38.3 (t), 41.9 (q), 54.3 (s), 56.1 (t), 56.8 (q), 66.2 (t), 110.0 (d), 110.6 (d), 112.9 (d), 133.6 (s), 137.4 (s), 156.5 (s), 182.8 (s); mlz (CI) 233 [(M+H)+), 100%].

Benzyl(2-iodophenyl)amine 13b

12b (5 g, 22.80 mmol, 1.00 equiv.), was treated as for **12a** and afforded benzyl(2-iodophenyl)amine **13b** as a yellow oil (6.5 g, 92%) [Found (ES): MH $^+$ 310.0095, C₁₃H₁₃IN requires MH 310.0093]; $\nu_{\rm max}$ (film)/cm $^{-1}$ 3400, 3060, 3027, 2854, 1589, 1502, 1318, 1004; $\delta_{\rm H}$ 4.52 (2H, d, J 5.6, CH₂), 4.74 (1H, br s, NH) 6.56 (1H, ddd, J 1.4, 7.3, 7.8, ArH), 6.64 (1H, dd, J 1.3, 8.1, ArH), 7.26 (1H, ddd, J 1.4, 7.3, 8.1, ArH), 7.39–7.48 (5H, m, ArH), 7.78–7.80 (1H, dd, J 1.4, 7.8, ArH); $\delta_{\rm C}$ 48.5 (t), 85.5 (s), 111.1 (d), 119.0 (d), 127.4 (d), 127.5 (d), 128.9 (d), 129.6 (d), 138.8 (s), 139.2 (d), 147.3 (s); m/z (EI) 309 [(MH) $^+$, 18%], 180 (20), 91 (100) and 65 (22).

3-[Benzyl(2-iodophenyl)carbamoyl]but-3-enoic acid methyl ester 15b

Benzyl(2-iodophenyl)amine **13b** (4.5 g, 14.55 mmol, 1.00 equiv.) was reacted as for **13a** and gave *3-[benzyl(2-iodophenyl)carbamoyl]but-3-enoic acid methyl ester* **15b** as a white solid mp 60–62 °C (5.96 g, 94%); [Found (EI): M+ 435.0339, $C_{19}H_{18}INO_3$ requires M 435.0331]; ν_{max} (film)/cm⁻¹ 2949, 1736, 1653, 1625, 1470, 1172; δ_{H} 3.06 (1H, d, J 16.5, CH₂), 3.67–3.70 (4H, m, CH₂ and CH₃), 4.23 (1H, d, J 14.3, CH₂), 5.15 (1H, s, CH₂), 5.27 (1H, s, CH₂), 5.78 (1H, d, J 14.3, CH₂), 6.94–7.00 (2H, m, ArH), 7.16–7.20 (1H, m, ArH), 7.24–7.30 (5H, m, ArH), 7.91 (1H, d, J 7.9, ArH); δ_{C} 40.0 (t), 52.0 (q), 52.5 (t), 99.8 (s), 123.4 (t), 127.7 (d), 128.4 (d), 129.0 (d), 129.4 (d), 129.6 (d), 132.0 (d), 136.4 (s), 136.6 (s), 140.2 (d), 144.7 (s), 168.9 (s), 171.3 (s); mlz (EI) 435 (M+, 5%), 376 (9), 308 (18), 180 (19) and 91 (100).

N-Benzyl-2-(2-hydroxyethyl)-N-(2-iodophenyl)acrylamide 16b

3-[Benzyl(2-iodophenyl)carbamoyl]but-3-enoic acid methyl ester **15b** (12.1 g, 27.8 mmol, 1.00 equiv.) was treated as for **15a** to afford *N-benzyl-2-(2-hydroxyethyl)-N-(2-iodophenyl)-acrylamide* **16b** as a colourless oil (9.5 g, 84%) [Found (ES): MH⁺ 408.0461, C₁₈H₁₉INO₂ requires *MH* 408.0460]; $\nu_{\rm max}$ (film)/cm⁻¹ 3401, 2931, 1647, 1619, 1469, 1030; $\delta_{\rm H}$ 2.27–2.31 (1H, m, CH₂), 2.50–2.57 (1H, m, CH₂), 3.20 (1H, br s, OH), 3.74–3.81 (2H, m, CH₂), 4.14 (1H, d, *J* 14.3, CH₂), 5.10 (1H, s, CH₂), 5.13 (1H, s, CH₂), 5.70 (1H, d, *J* 14.3, CH₂), 6.68 (1H, d, *J* 7.6, ArH), 6.97–7.00 (1H, m, ArH), 7.15–7.28 (6H, m, ArH), 7.92 (1H, dd, *J* 1.4, 7.9, ArH); $\delta_{\rm C}$ 38.3 (t), 52.2 (t), 62.8 (t), 99.9 (s), 120.7 (t), 127.9 (d), 128.7 (d), 128.9 (d), 129.6 (d), 129.7 (d), 131.6 (d), 136.6 (s), 140.4 (d), 141.9 (s), 144.4 (s), 171.9 (s); *m/z* (EI) 407 (2%, M⁺), 376 (12), 280 (17), 180 (17), 91 (100) and 65 (25).

2-(2-Azidoethyl)-N-benzyl-N-(2-iodophenyl)acrylamide 17b

Alcohol **16b** (480 mg, 1.18 mmol, 1.00 equiv.) was treated as for alcohol **16a** to afford pure 2-(2-azidoethyl)-N-benzyl-N-(2-iodophenyl) acrylamide **17b** as a slightly yellow oil (357 mg, 70%); [Found (EI): M⁺ 432.0446, C₁₈H₁₇IN₄O requires M 432.0447]; ν_{max} (film)/cm⁻¹ 3031, 2934, 2875, 2095, 1651, 1624, 1469, 1385; δ_{H} 2.37–2.41 (1H, m, CH₂), 2.57–2.64 (1H, m, CH₂), 3.44 (2H, t, J 6.7, CH₂), 4.16 (1H, d, J 14.2, CH₂), 5.12 (1H, s, CH₂), 5.17 (1H, s, CH₂), 5.70 (1H, d, J 14.2, CH₂), 6.70 (1H, d, J 7.5, ArH), 6.96–7.00 (1H, m, ArH), 7.16–7.38 (6H, m, ArH),

7.90 (1H, dd, J 1.4, 7.8, ArH); $\delta_{\rm C}$ 33.9 (t), 50.1 (t), 52.2 (t), 99.9 (s), 120.3 (d), 120.7 (t), 125.8 (d), 127.9 (d), 128.6 (d), 129.0 (d), 129.6 (d), 130.1 (d), 131.7 (d), 136.9 (s), 140.5 (d), 140.5 (s), 145.4 (s) 170.5 (s); m/z (CI) 432 (100%, M^+).

Tandem radical cyclisation of 2-(2-azidoethyl)-N-benzyl-N-(2-iodophenyl)acrylamide 17b

2-(2-Azidoethyl)-N-benzyl-N-(2-iodophenyl)acrylamide (160 mg, 0.37 mmol, 1.00 equiv.) when treated as for azide 17a gave pure 1-benzyl-1'-methyl-1H-spiro[indole-3,3'-pyrrolidin]-2-one 23b as a colourless oil (73.6 mg, 68% over two steps) [Found (ES): MH^+ 293.1652, $C_{19}H_{21}N_2O$ requires MH293.1654]; v_{max} (film)/cm⁻¹ 2940, 2838, 2786, 1711, 1611, 1486, 1467, 1359, 1183; $\delta_{\rm H}$ 2.11–2.18 (1H, m, CH₂), 2.41–2.46 (1H, m, CH₂), 2.50 (3H, s, CH₃), 2.77–2.83 (1H, m, CH₂), 2.89 (1H, d, J 9.4, CH₂), 2.95 (1H, d, J 9.4, CH₂), 3.09–3.14 (1H, m, CH₂), 4.9 (2H, s, CH₂), 6.71 (1H, d, J7.7, ArH), 7.04 (1H, ddd, J1.5, 7.3, 7.7, ArH), 7.14 (1H, ddd, J 0.8, 7.7, 7.7, ArH), 7.24–7.33 (5H, m, ArH), 7.44 (1H, dd, J 1.4, 7.8, ArH); $\delta_{\rm C}$ 38.3 (t), 42.0 (q), 44.0 (t), 53.5 (s), 56.9 (t), 66.6 (t), 108.9 (d), 123.1 (d), 123.2 (d), 127.4 (d), 127.7 (d), 127.8 (d), 129.0 (d), 136.0 (s), 136.2 (s), 142.1 (s), 180.6 (s); m/z (EI) 292 [(M)⁺, 30%], 235 (100), 91(80).

(±)-Coerulescine 7

1-Benzyl-1'-methyl-1*H*-spiro[indole-3,3'-pyrrolidin]-2-one **23b** (22 mg, 0.075 mmol, 1.00 equiv.) was treated as for **23a** and afforded pure (±)-coerulescine 7 as a pale yellow gum (12.7 mg, 83%); [Found (ES): MH⁺ 203.1182, $C_{12}H_{14}N_2O$ requires *MH* 203.1184]; v_{max} (film)/cm⁻¹ 3385, 2969, 2793, 1711, 1619, 1471, 1336, 1196, 1047; δ_{H} 2.10–2.18 (1H, m, CH₂), 2.39–2.45 (1H, m, CH₂), 2.50 (3H, s, CH₃), 2.80–2.86 (1H, m, CH₂), 2.89–2.96 (2H, m, CH₂), 3.05–3.11 (1H, m, CH₂), 6.88 (1H, d, *J* 7.7, ArH), 7.07 (1H, ddd, *J* 0.9, 7.2, 7.7, ArH), 7.20 (1H, ddd, *J* 1.2, 7.7, 7.7, ArH), 7.42 (1H, d, *J* 7.2, ArH), 8.16 (1H, br s, NH); δ_{C} 38.1 (t), 41.9 (q), 53.9 (s), 56.9 (t), 66.2 (t), 109.8 (d), 123.1 (d), 123.5 (d), 128.0 (d), 136.1 (s), 140.4 (s), 183.3 (s); m/z (EI) 202 (40%, M⁺), 145 (25), 130 (28), 57 (100) and 51 (12).

Acknowledgements

We thank: the Royal Society for a Leverhulme Senior Research Fellowship to JAM; University of Strathclyde for a studentship to D. L. and EPSRC Mass Spectrometry Service Swansea for mass spectra.

References

- (a) C.-B. Cui, H. Kakeya, G. Okada, R. Onose and H. Osada,
 J. Antibiot., 1996, 49, 527; (b) C.-B. Cui, H. Kakeya and H. Osada,
 J. Antibiot., 1996, 49, 832; (c) C.-B. Cui, H. Kakeya and H. Osada,
 Tetrahedron. 1996, 52, 12651.
- (a) P. R. Sebahar and R. M. Williams, J. Am. Chem. Soc., 2000, 122, 5666; (b) H. Wang and A. Ganesan, J. Org. Chem., 2000, 65, 4685; (c) L. E. Overman and M. D. Rosen, Angew. Chem., Int. Ed., 2000, 39, 4596; (d) T. D. Bagul, G. Lakshmaiah, T. Kawabata and K. Fuji, Org. Lett., 2002, 4, 249; (e) T. Lindel, Nachr. Chem., 2000, 48, 1498; (f) H. S. Wang and A. Ganesan, J. Org. Chem., 2000, 65, 4685; (g) F. von Nussbaum and S. J. Danishefsky, Angew. Chem., Int. Ed., 2000, 39, 2175.
- 3 S. Edmondson, S. J. Danishefsky, L. Sepp-Lorenzino and N. Rosen, J. Am. Chem. Soc., 1999, 121, 2147.
- 4 (a) A. Jossang, P. Jossang, H. A. Hadi, T. Sévenet and B. Bodo, J. Org. Chem., 1991, 56, 6527; (b) K. Jones and J. Wilkinson, J. Chem. Soc., Chem. Commun., 1992, 1767; (c) C. Pellegrini, C. Strässler, M. Weber and H.-J. Borschberg, Tetrahedron: Asymmetry, 1994, 5, 1979; (d) S.-I Bascop, J. Sapi, J.-Y. Laronze and J. Lévy, Heterocycles, 1994, 38, 725; (e) G. Palmisano, R. Annunziata, G. Papeo and M. Sisti, Tetrahedron: Asymmetry, 1996, 7, 1; (f) G. Lakshmaiah, T. Kawabata, M. Shang and K. Fuji, J. Org. Chem., 1999, 64, 1699; (g) C. Fischer, C. Meyers and E. M. Carreira, Helv. Chim. Acta, 2000, 83, 1175; (h) G. Cravotto, G. B. Giovenzana, T. Pilati, M. Sisti and G. Palmisano, J. Org. Chem., 2001, 66, 8447; (i) M. E. Kuehne, D. M. Roland and R. Hafter, J. Org. Chem., 1978, 43, 3705; (j) U. K. Syam Kumar, H. Ila and H. Junjappa, Org. Lett., 2001, 3, 4193; (k) M. Sonei, K. Noguchi, R. Yamagami, Y. Kawada, K. Yamada and F. Yamada, Heterocycles, 2000, 53, 7; (l) S. E. V. Bell, R. F. C. Brown, F. W. Eastwood and J. M. Horvath, Aust. J. Chem., 2000, 53, 183.
- N. Anderton, P. A. Cockrum, S. M. Colegate, J. A. Edgar, K. Flower, I. Vit and R. I. Willing, *Phytochemistry*, 1998, 48, 437.
- 6 B. Patro and J. A. Murphy, Org. Lett., 2000, 2, 3599.
- 7 S.-Z. Zhou, S. Bommezijn and J. A. Murphy, Org. Lett., 2002, 4, 443.
- 8 For preliminary report, see: D. Lizos, R. Tripoli and J. A. Murphy, *Chem. Commun.*, 2001, 2732.
- 9 Y. Kondo, S. Kojima and T. Sakamoto, *J. Org. Chem.*, 1997, **62**, 6507.
- 10 L. F. Tietze and T. Grote, Chem. Ber., 1993, 126, 2733.
- 11 B. R. Baker, R. E. Schaub and J. H. Williams, J. Org. Chem., 1952, 17, 116.
- 12 S. E. Webber, K. Okano, T. L. Little, S. H. Reich, Y. Xin, S. A. Fuhrman, D. A. Matthews, R. A. Love, T. F. Hendrickson, A. K. Patick, J. W. Meador, R. A. Ferre, E. L. Brown, C. E. Ford, S. L. Binford and S. T. Worland, J. Med. Chem., 1998, 41, 2786 note that commercial lithium borohydride was not tried in this experiment.
- 13 K. Jones, S. A. Brunton and R. Gosain, Tetrahedron Lett., 1999, 40, 8935
- 14 A. R. Lee, W. H. Huang, T. L. Lin, K. M. Shih and H. F. Lee, J. Heterocycl. Chem., 1995, 32, 1.